

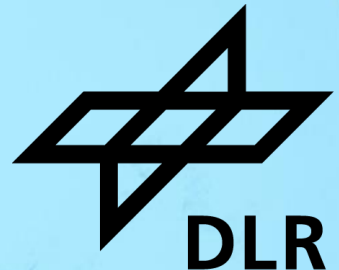


FROM STORAGE TO STABILITY: TECHNOLOGIES FOR TOMORROW'S GRID HYDROGEN

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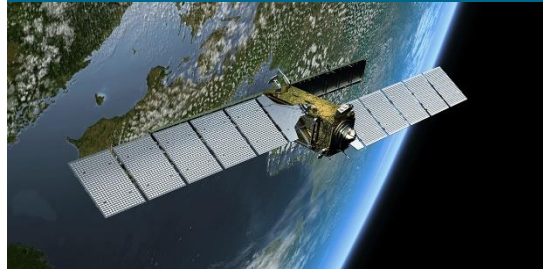
DLR - Research Organisation + Space Agency + Project Management Agency



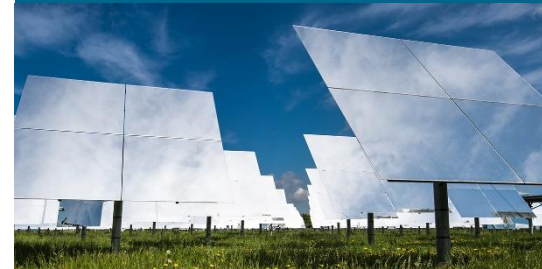
Aeronautics



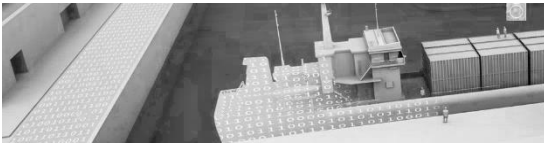
Aerospace



Energy

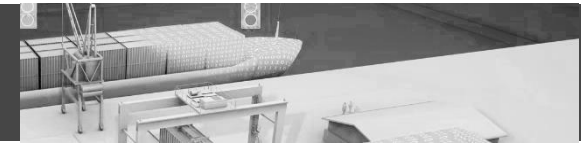


Transport



Security

security research (civil & defense)



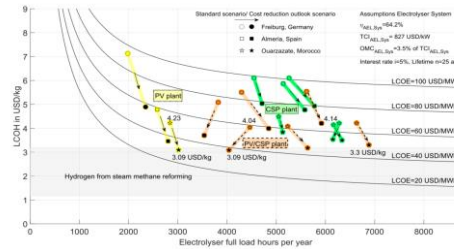
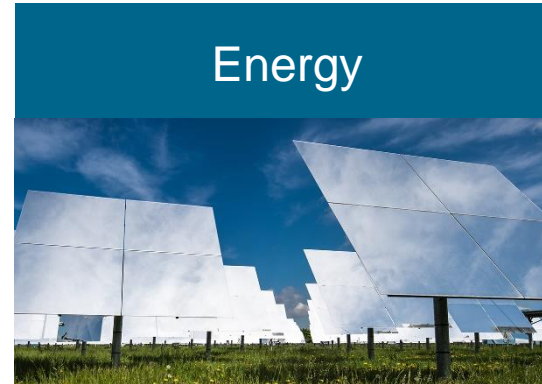
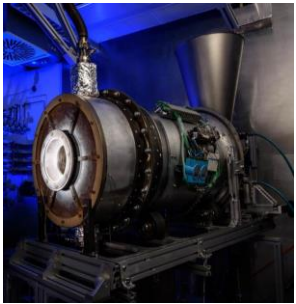
Digitalisation

Quantum technologies & system modelling



- Largest European research organisation for engineering and aerospace
- Close cooperation with research, economy and industry actors
- Institutionally funded by Federal Ministry for Economic Affairs and Climate Action, additionally by Federal Ministry of Defence, project funding by many others (e. g. Federal Ministry of Education and Research)

DLR - Research Organisation



Institute of Future Fuels (FF)



- Technology development and assessment for efficient and economical production of energy sources for a global, renewable energy economy
- Contributions to the decarbonization of energy, aviation and transport

Motivation: Hydrogen as an energy carrier



- **Hydrogen:**

- Not primary source energy like sunlight but energy carrier
- Energy density 1kg of hydrogen contains as much energy as 2.1 kg of natural gas or 2.8 kg of petrol

- **Challenges:**

- Climate neutrality
- How can stable year-round system operation with renewable energy be guaranteed while minimising overall costs
- How can non-electrifiable industries become carbon neutral

Motivation: Hydrogen as an energy carrier

■ Hydrogen:

- Key role as energy carrier for the transport sector
- Industrial processes (e.g. production of fertilisers)
- As energy buffer for integration of large amounts of renewable energy into the grid
 - Goal of climate neutrality requires a massive expansion of renewable electricity generation, especially wind and PV
 - Fluctuating renewable electricity is becoming the most important primary energy source
 - Need for long-term storage to bridge dark doldrums



Refining



Ammonia / Fertilizer



Transportation



Chemicals



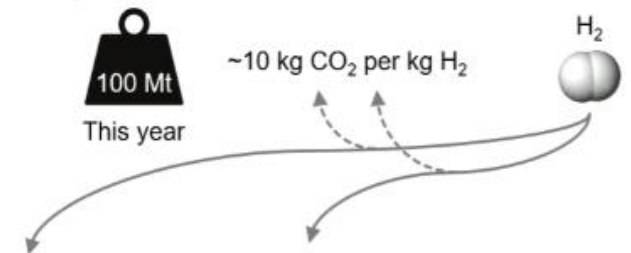
Steelmaking

Motivation: Hydrogen as an energy carrier

■ Hydrogen Market:

Year	1975	1990	2000	2010	2018	2021	2025	2030*	2040
H ₂ consumption [Mio. Ton]	18,2	34	52,4	62	73,9	94	100	>140	~200

- Over 96% of current hydrogen production: fossil fuels with significant associated CO₂-emissions
 - Steam methane reforming (SMR): most common way to produce hydrogen
 - Oil and coal gasification: also widely used (China and Australia)

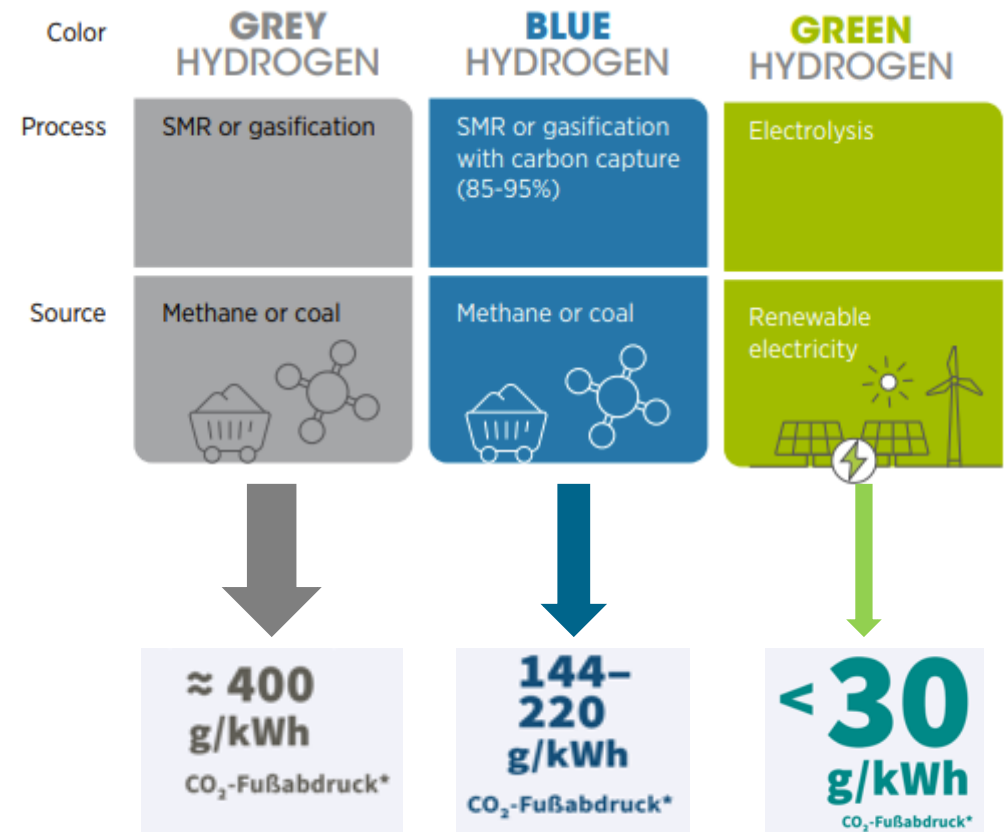


- 2023: associated emissions of about 920 Mt CO₂

➡ Development of carbon lean technologies producing hydrogen from renewable sources: of utmost importance

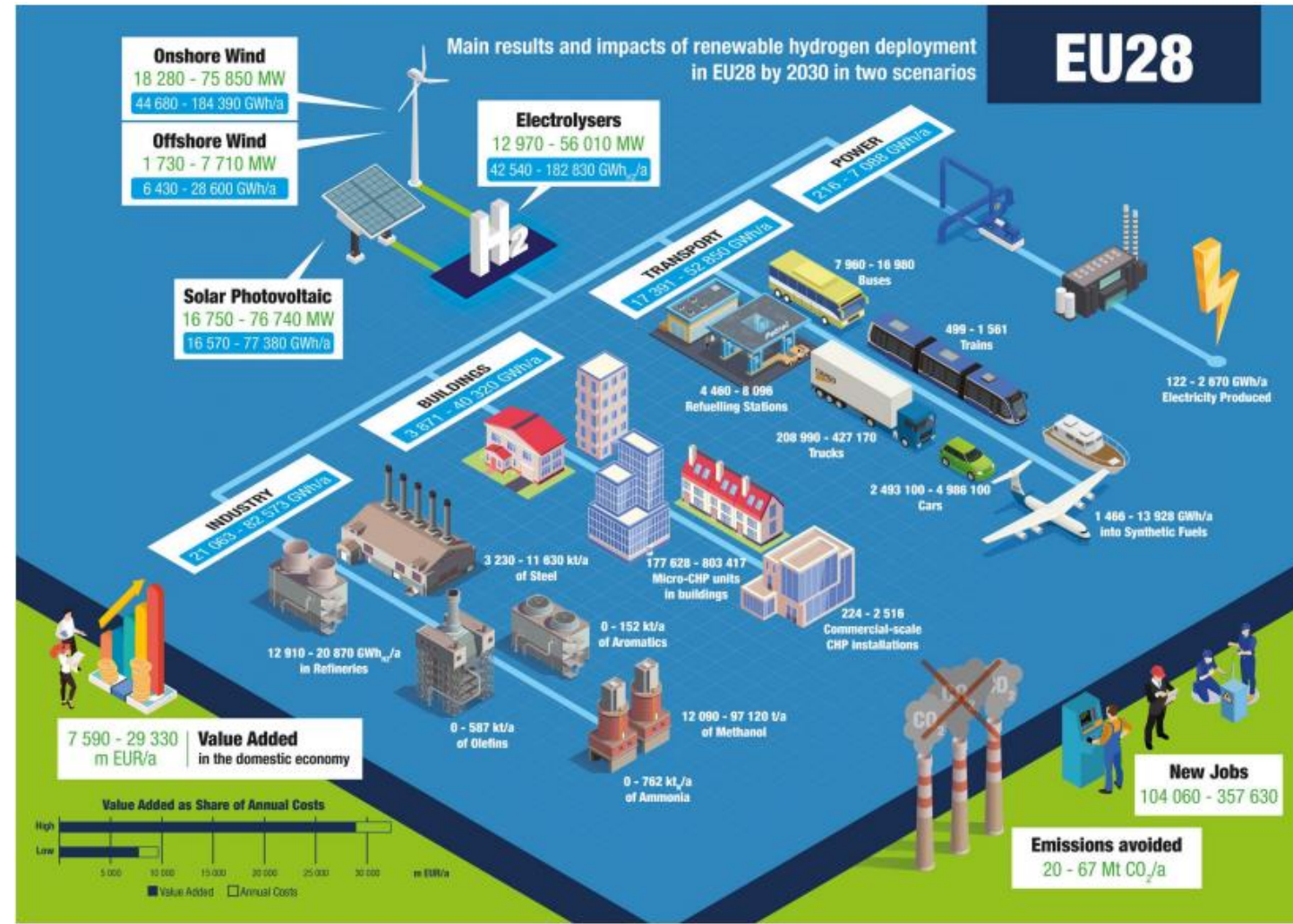
CO₂ emissions from Hydrogen

- **Grey hydrogen:** produced from methane (CH₄), split with steam into CO₂ – the main culprit for climate change – and H₂. Grey hydrogen has increasingly been produced also from coal, with significantly higher CO₂ emissions.
- **Blue hydrogen:** same process as grey, with the additional technologies necessary to capture the CO₂ produced when hydrogen is split from methane (or from coal) and store it for long term.
- **Green hydrogen:** produced by splitting water into hydrogen and oxygen using renewable energy.

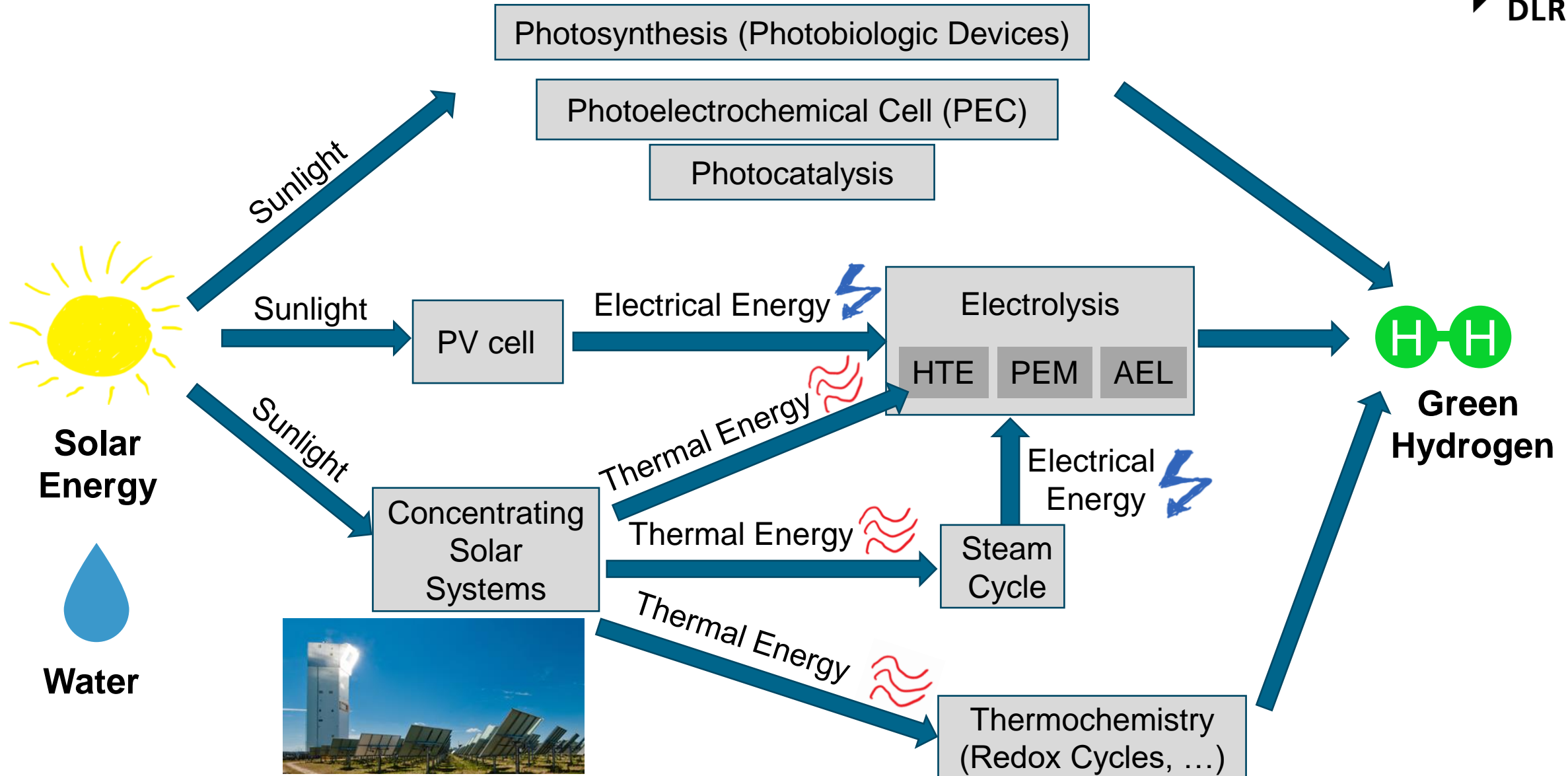


Hydrogen – Application Spectrum

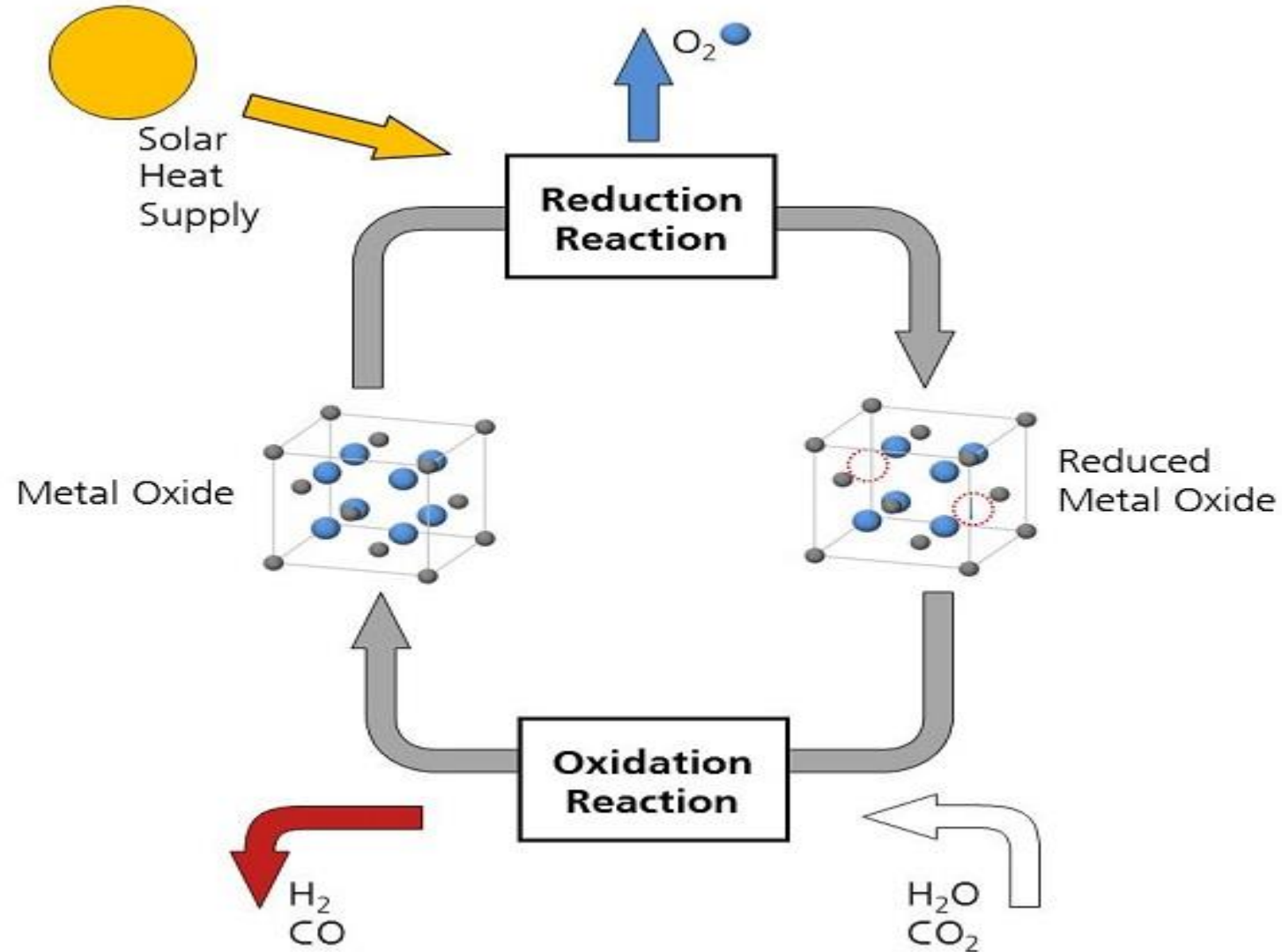
- Goal: fast and sustainable climate neutrality
 - The variety of water splitting technologies are not known enough
- Green hydrogen – will play an important role in the changing energy system and in the **changing chemical industry**
- Hydrogen will be used as
 - Energy vector (fuel)
 - Reducing agent (iron ...)
 - Building block (ammonia ...)



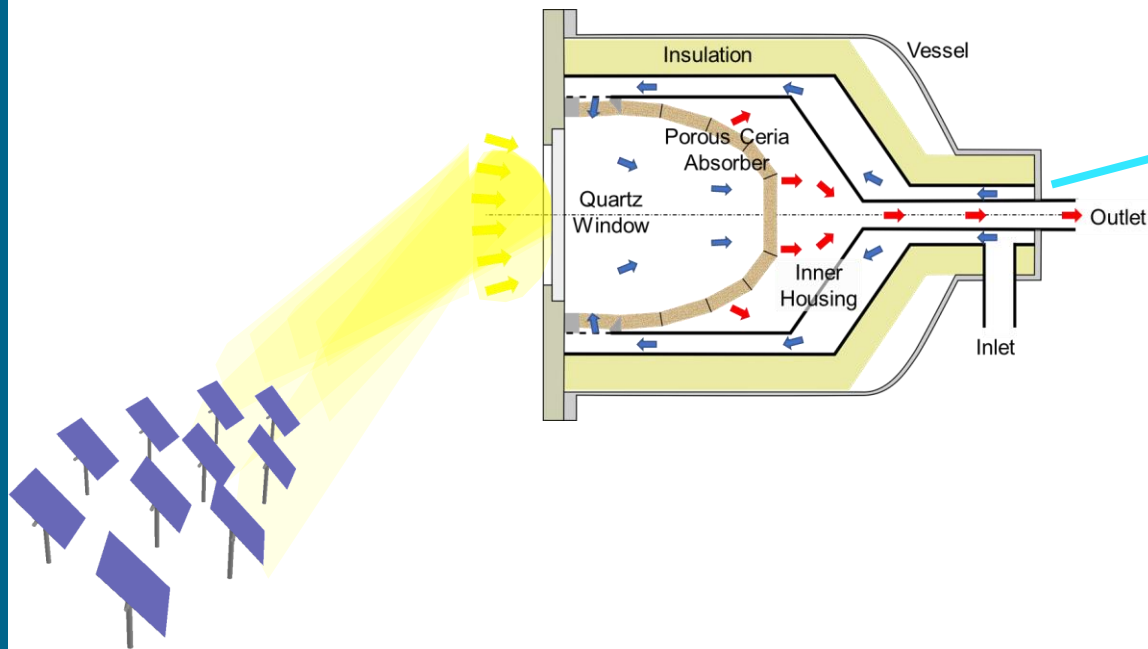
Green Hydrogen from Water Splitting by Solar Energy



Schematic of the two-stage cycle with metal oxides



Schematic of the two-stage cycle with metal oxides



Heliostat Field

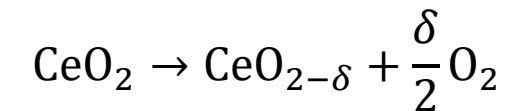


MO = ferrites, ceria, perovskites...

Long thermal stability
High O₂ storage capacity
Lower sintering
High melting point
Good redox reactivity

- Various metal oxide redox pairs
- Reaction with cerium oxide

1st step: Reduction



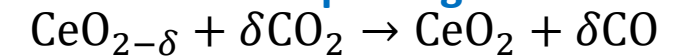
1500 °C, reduced p_{O₂}

2nd step: Oxidation

Water splitting



CO₂ splitting



900 °C, atmospheric pressure

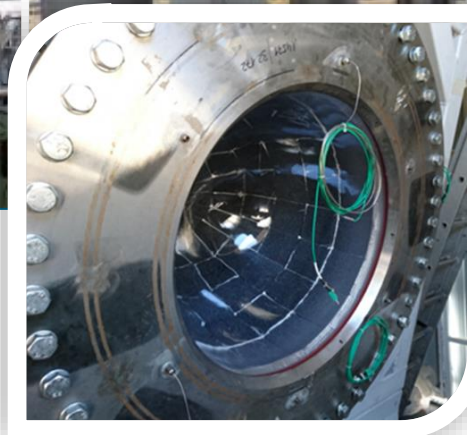
Solar Thermochemical Hydrogen Production

State of the Art: Batch Reactors with Fixed, Porous Monoliths of Redox Material

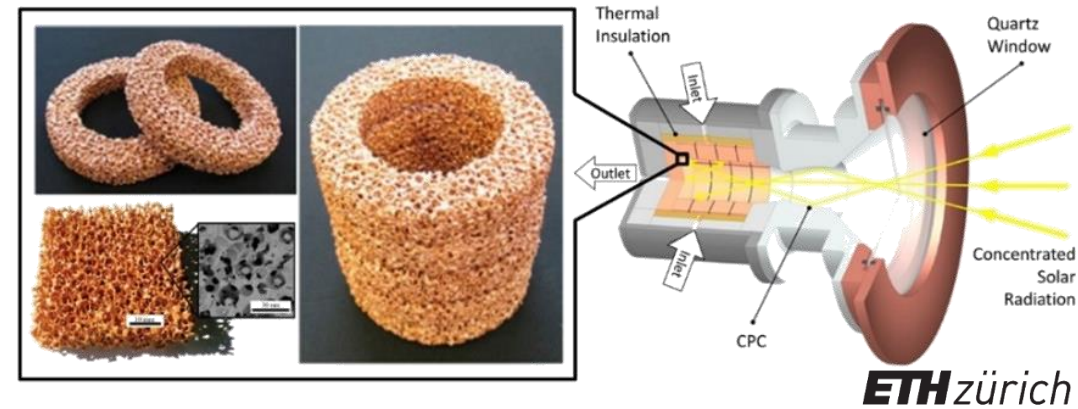


Source: Plataforma Solar de Almería
(Owned by the Spanish research
centre CIEMAT)

Source: DLR



Source: IMDEA / SUN-to-LIQUID project



■ **Hydrogen:**

- Key role: energy carrier for the transport sector, industrial processes, energy buffer
- Could become one of the most powerful fuel of our lifetime

■ **Green Hydrogen:**

- One of the solutions for an energy transition toward zero or net-zero emissions
- Current wave focuses on creating a link between renewable electricity and hard-to-electrify end uses
- Different pathways of production
- National pledges to achieve net-zero emissions
- More extensive base of interested stakeholders



synlight

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Gefördert durch:

Ministerium für Klimaschutz, Umwelt,
Landwirtschaft, Natur- und Verbraucherschutz
des Landes Nordrhein-Westfalen



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Bundesministerium
für Wirtschaft
und Energie

aufgrund eines Beschlusses
des Deutschen Bundestages



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Thank you for your attention!